

Spatially Distributed Experimentation to Understand ALD Process Chemistry

Rubloff Research Group Accomplishments

Spatially Distributed Atomic Layer Deposition (ALD)

Accomplishment

Designed and implemented ALD reactor to achieve spatial distribution of impingement fluxes

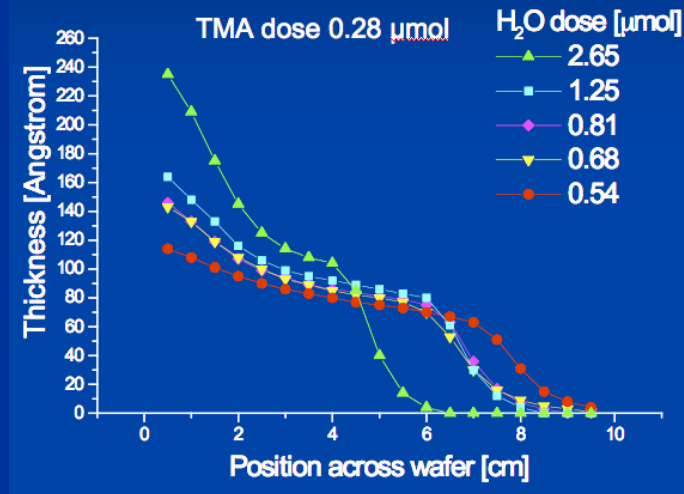
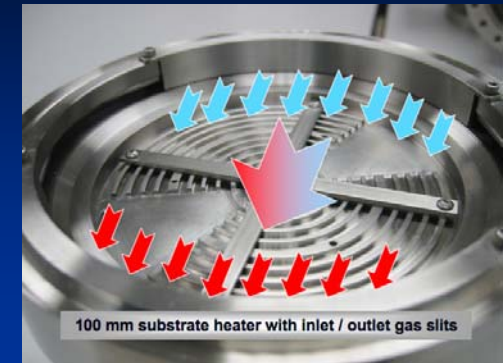
Demonstrated spatial gradients in thickness and material properties

Significance

Atomic layer deposition (ALD) is widely sought for its atomic-scale thickness control and unprecedented uniformity and conformality

Precursor dose permutations add complexity to ALD process recipes, where underlying surface chemistry is not well understood

Spatially distributed ALD enables rapid investigation of ALD process/dose recipes and optimization of material properties



Researchers involved

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Support

NSF IMI, PAT-IRST MicroCombi, Laboratory for Physical Sciences, MKS Instruments

Spatially Distributed Atomic Layer Deposition (ALD)

Intellectual merit

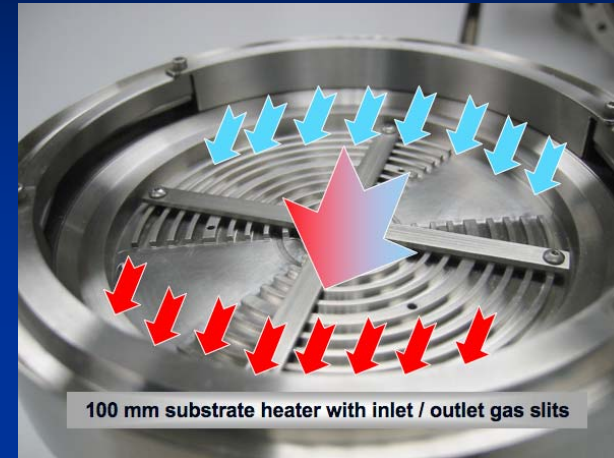
Atomic layer deposition (ALD) enables atomic-level control of ultrathin film deposition over 3-D surfaces. It relies on alternating doses of molecules that saturate surface sites to achieve these benefits.

ALD surface chemistry is not well-understood, so finding suitable dose recipes is hampered by the numerous permutations available.

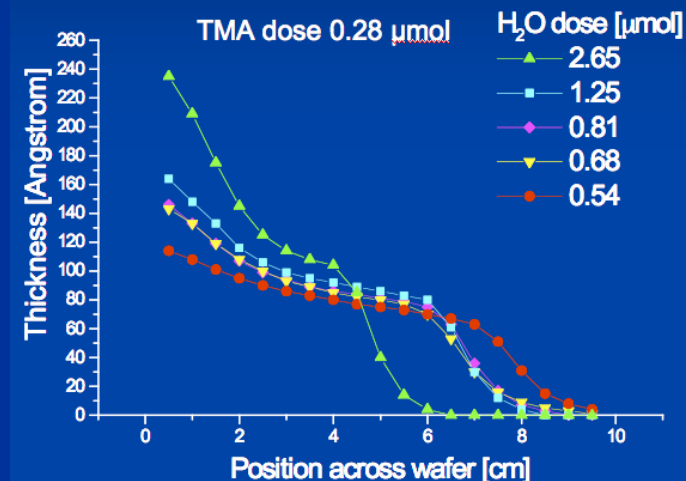
We have designed and implemented an ALD reactor design that uses cross-flow of precursor gases to achieve combinatorial variation of surface reaction conditions across a 4" wafer.

We have demonstrated the fabrication of spatial gradients in thickness and material properties for an Al_2O_3 ALD process.

Coupled with rapid characterization of across-wafer material and electrical properties, the cross-flow ALD reactor provides a platform for addressing the relation of ALD process chemical conditions to resulting material properties.



100 mm substrate heater with inlet / outlet gas slits



Spatially Distributed Atomic Layer Deposition (ALD)

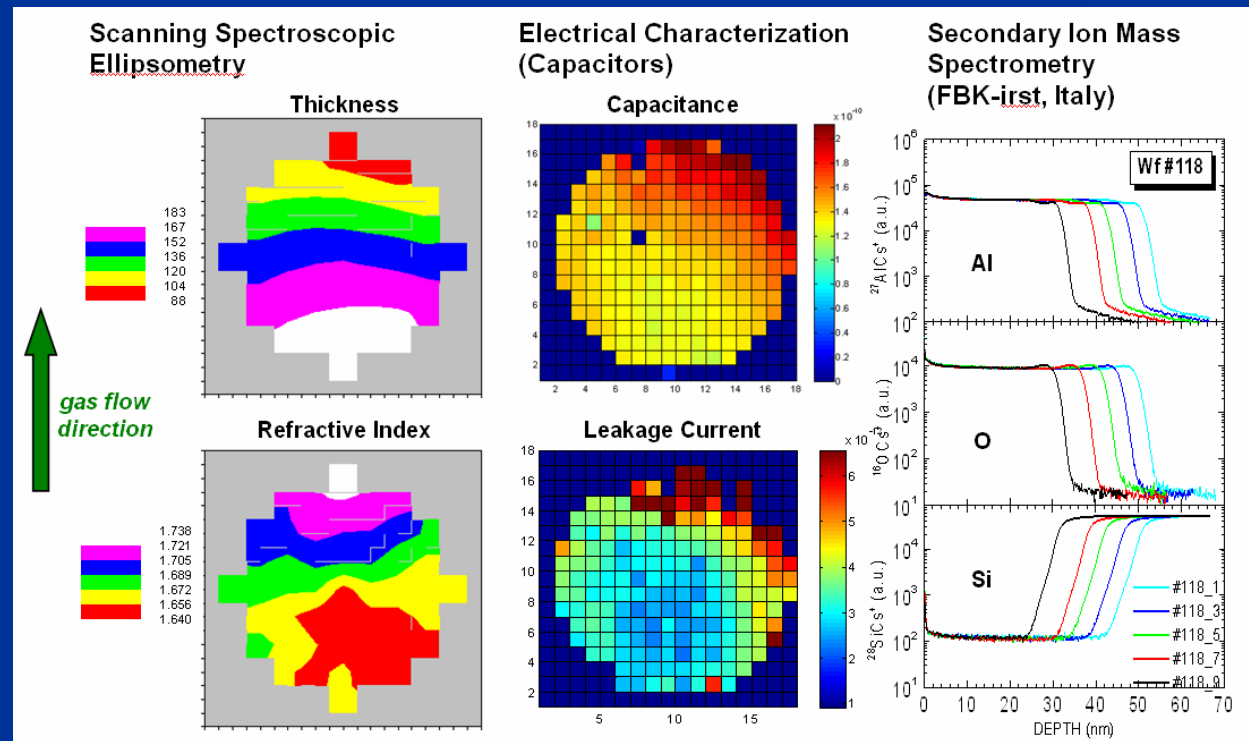
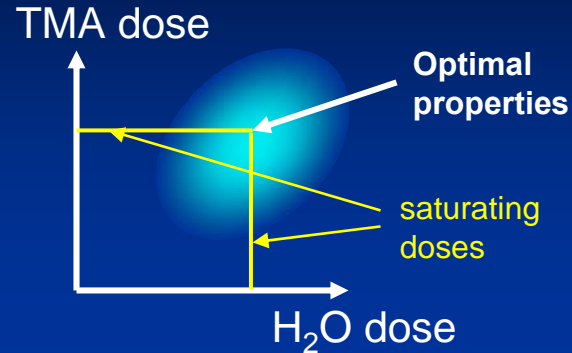
Broader Impacts

Atomic layer deposition (ALD) is widely sought for its atomic-scale thickness control and unprecedented uniformity and conformality, key benefits for existing technologies (e.g. semiconductors) and for a broad set of nanotechnology applications.

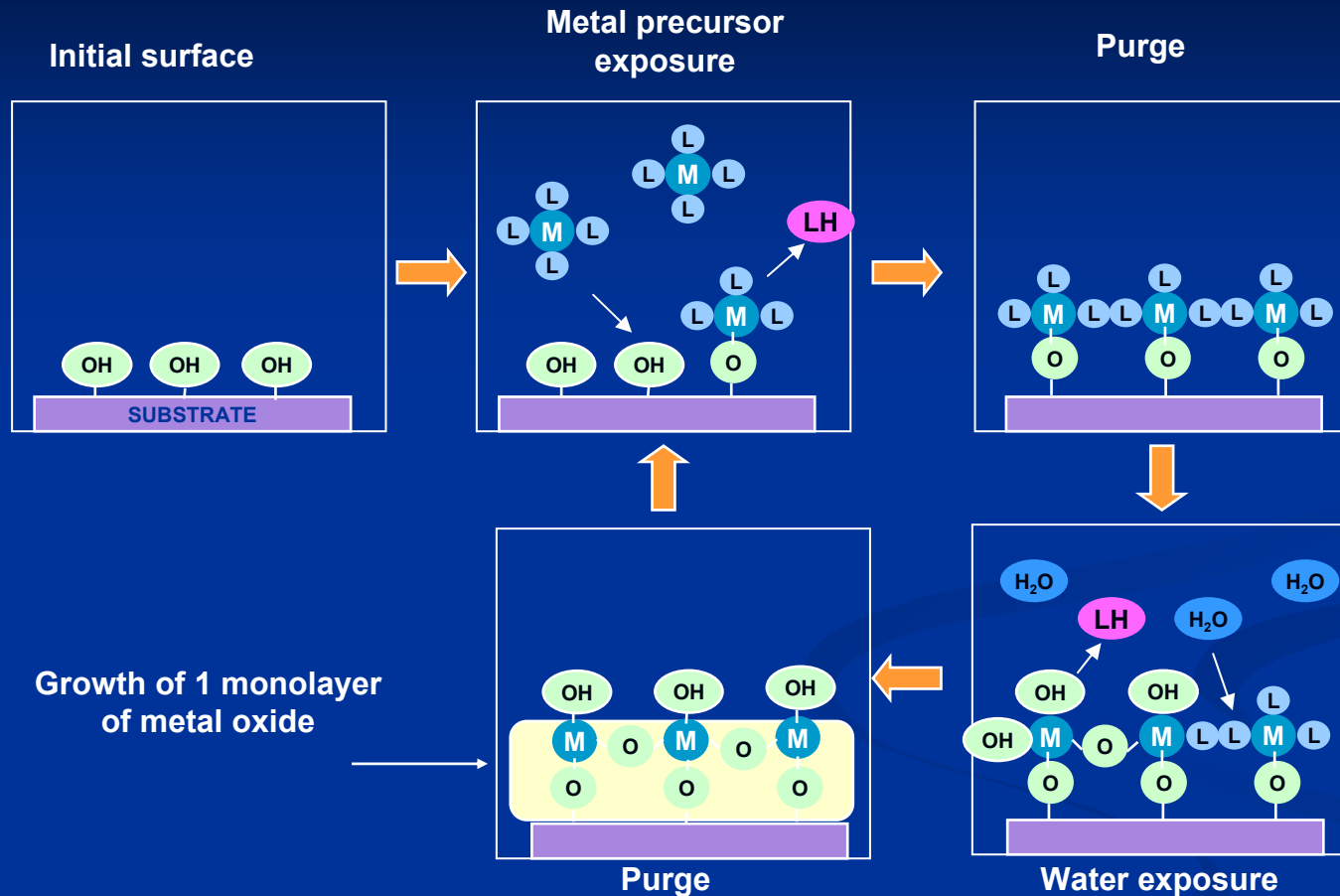
The numerous precursor dose permutations inhibits fundamental understanding of ALD surface chemistry and process applications.

Our cross-flow ALD design and associated material characterization provide a platform for rapid learning in ALD process behavior and enhanced development of ALD applications.

Close collaboration with Italy group (FBK-irst) has been essential in chemical, compositional, and structural characterization of ALD layers.

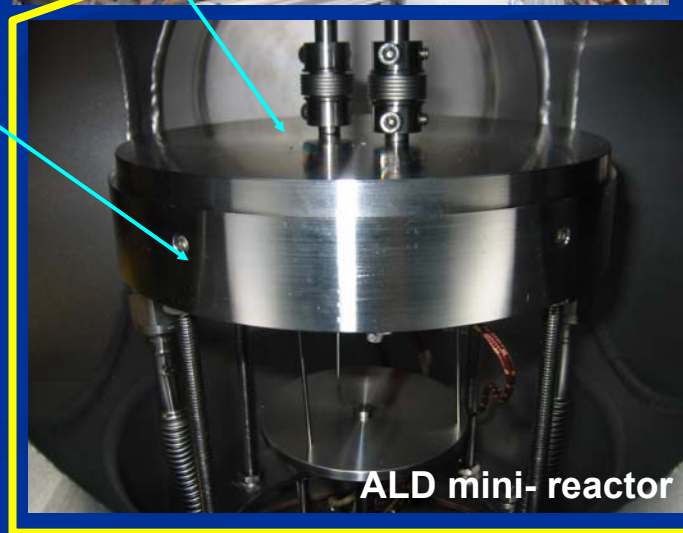
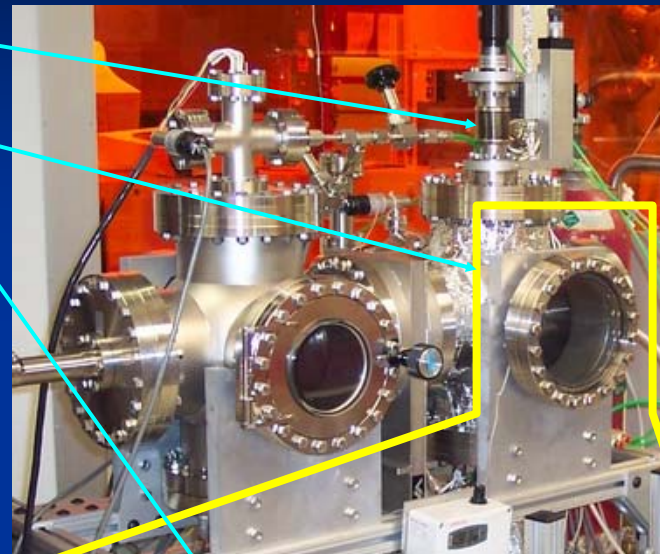
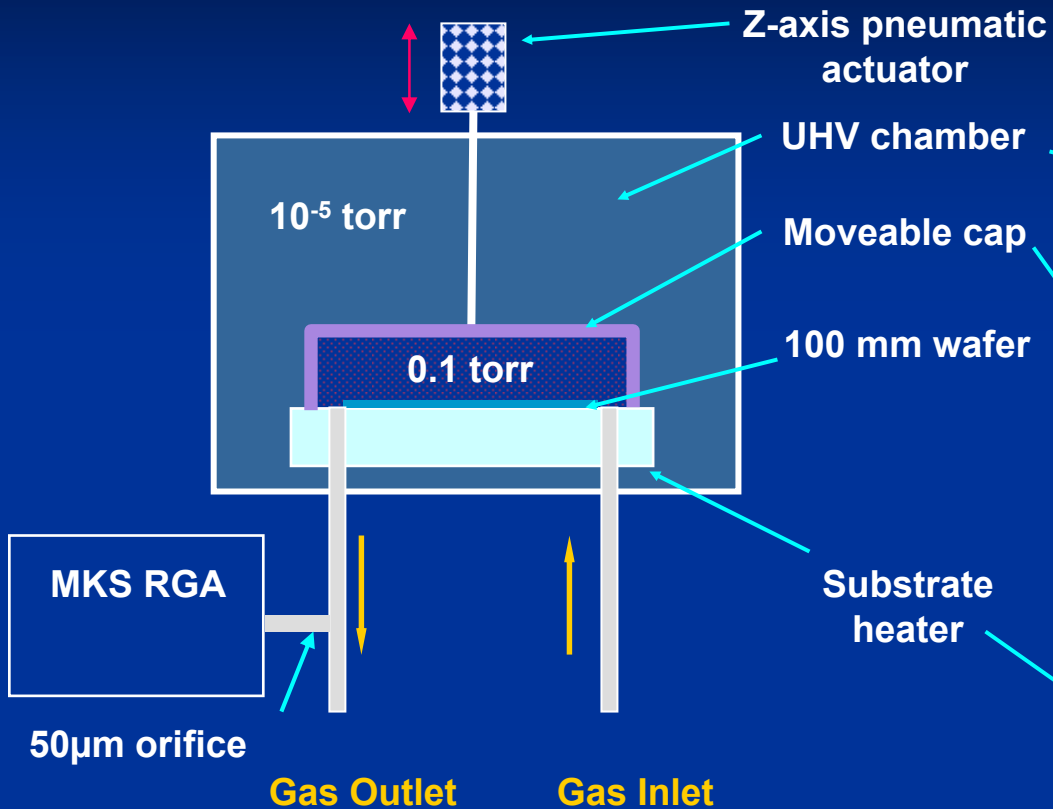


Atomic Layer Deposition (ALD)



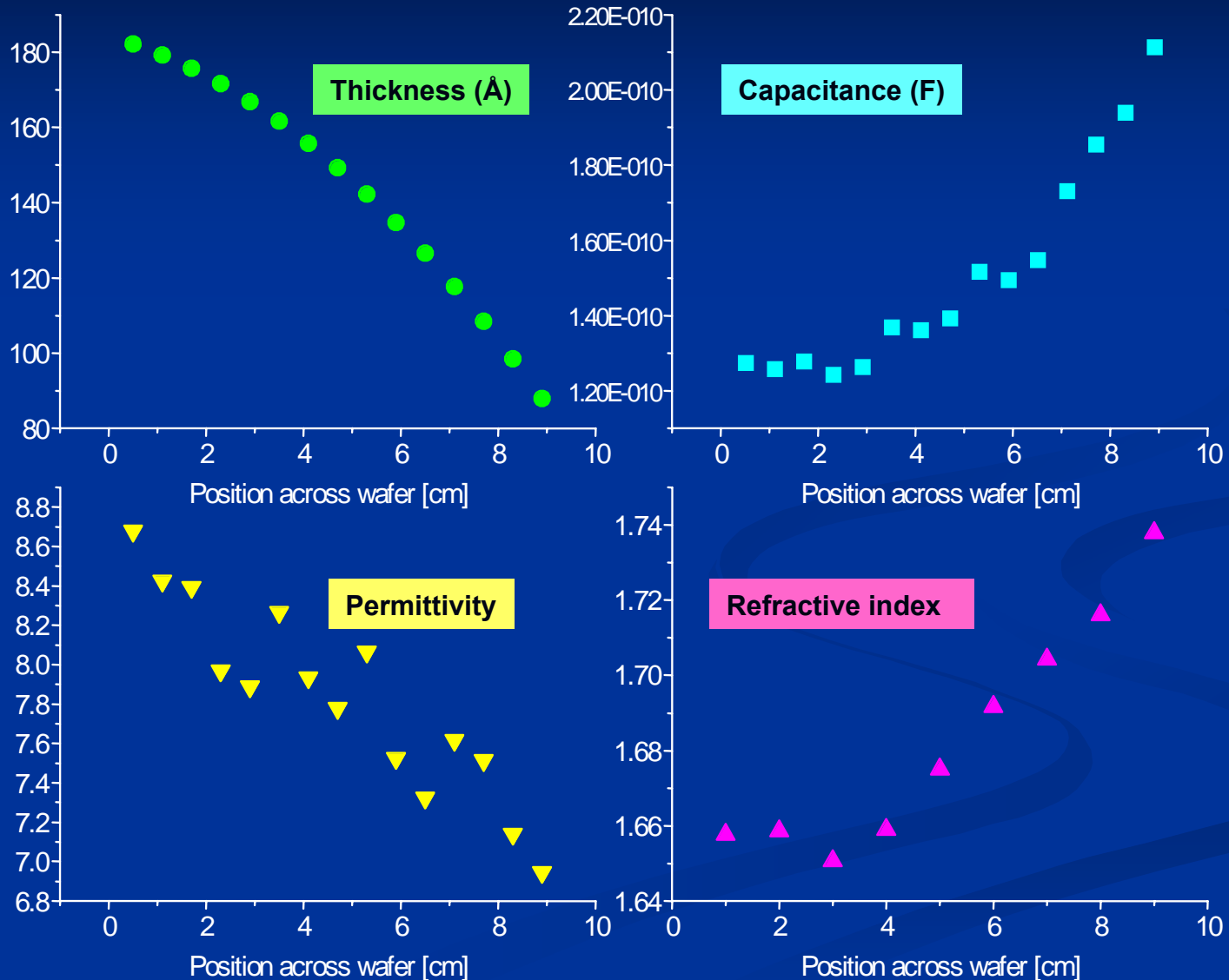
- **Alternating doses of chemical precursors, repeated approximately once per atomic layer**
- **Self-limiting adsorption/reaction for each dose**

Cross-Flow ALD Reactor



Across-Wafer Al_2O_3 ALD Film Properties

Under-dose H_2O conditions ($0.8\mu\text{mol}$)



Al₂O₃ ALD: TMA under-dosing

Moderate under-dosing

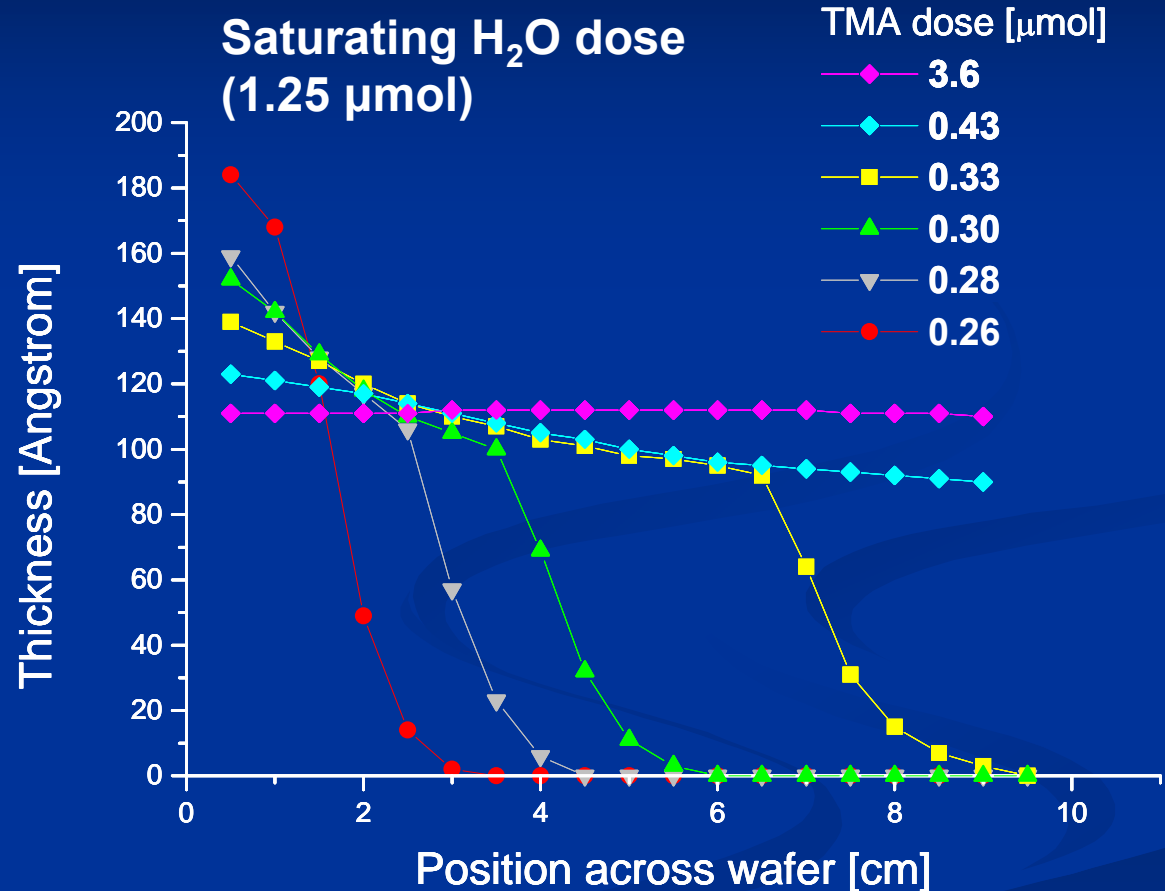
TMA causes uniformity degradation & depletion

Larger under-dosing

TMA condition generates very strong nonuniformity, nearly discontinuous profile

Enhance growth rate at inlet with dramatic depletion at outlet

→ Possibly OH oversaturation at inlet which getters and reacts the under-dosed TMA



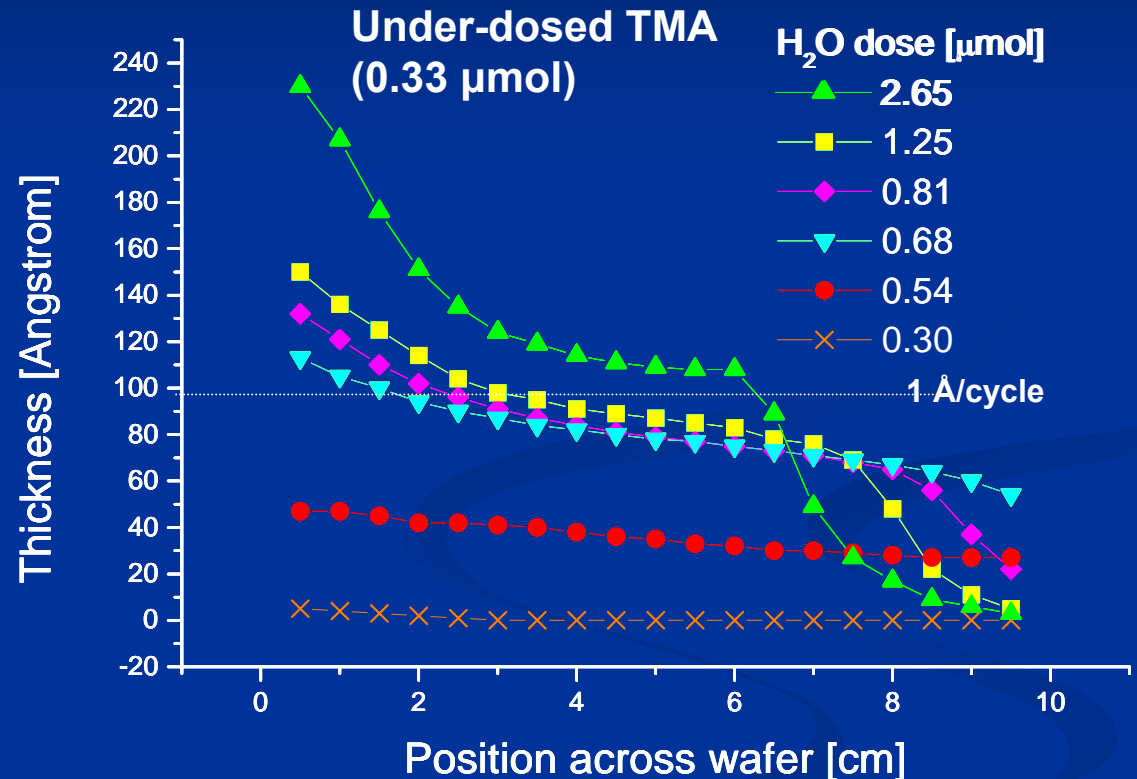
Al₂O₃ ALD: H₂O under-dosing

H₂O/TMA < 2

→ improved
uniformity but mean
thickness
decreases due to
insufficient reactant
supply

H₂O/TMA > 2

→ profound
nonuniformity at
front and rear, with
limited uniformity at
middle



9/6/2007, graph 2